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SYNTHESIS OF NOVEL TAXOL ANALOGS AND EVALUATION OF THEIR BIOLOGICAL ACTIVITIES

Paul A. Wender*, Daesung Lee and Tapan K. Lal Department of Chemistry, Stanford University, Stanford, CA 94305

Susan B. Horwitz* and Srinivasa Rao
Department of Molecular Pharmacology, Albert Einstein College of Medicine, Bronx, NY 10461

Abstract: Two new taxol analogs 6 and 10 have been prepared from baccatin III (1) and taxol (7a), respectively. Like taxol, both compounds were found to promote microtubule formation and stabilization, although they were less active than taxol. Both 6 and 10 exhibited cytotoxicity against J774.2 cells; 6 was ~60-fold less active and 10 was ~15-fold less active. © 1997 Elsevier Science Ltd.

Taxol^{®1} is a novel diterpenoid extracted from the bark of the western yew, *Taxus brevifolia*,² and a particularly significant new lead for the treatment of cancer.³ Thus far, it has been approved by the FDA for the treatment of advanced ovarian and metastatic breast cancer and is currently in phase II and III clinical trials for lung and other cancers.⁴ Adding to the medicinal significance of this compound is the finding that it operates through a novel mode of action involving facilitated assembly and stabilization of microtubules.⁵ While the molecular basis for this action is not understood, advances in recent years have done much to define which structural features of taxol are required for activity.⁶ Photoaffinity labeling studies with taxol analogs indicate that taxol binds to the N-terminal region of β-tubulin and a peptide containing β-tubulin amino acid residues 217-231.7

One of the principal goals associated with research in this area is the identification of the structural features of taxol required for its biochemical and physiological performance. Ultimately, such research can be expected to lead to the identification of structurally simpler and clinically more effective analogs that could be made in a practical fashion through total synthesis. Toward this end, we now report the synthesis and assay of the novel taxol analogs 6 and 10. The former is a representative member of a new analog class, the first to incorporate selenium and is readily derived from baccatin III (1). The latter is a novel C-ring contracted analog which has also been independently obtained from taxol (7a) by the Kingston group.

The synthesis of 6 started with baccatin III (1) (Scheme I). Reductive cleavage of the 10-acetoxy group was achieved with samarium diiodide and acetic acid in THF. ¹⁰ Treatment of the resultant product (2) with an excess amount (10–15 equiv) of selenium dioxide in pyridine at 85 °C yielded the selenophene-containing compound 3¹¹ in 45% yield together with 10-deacetoxy-13-oxo-baccatin III (4) (25%). Coupling of 3 with β-lactam 5¹² in the presence of sodium hexamethyldisilazide (2.5 equiv) in THF followed by deprotection with hydrogen fluoride in pyridine led to the selenophene-containing taxol analog 6 in 85% yield. The structure of 6 was assigned on the basis of ¹H, ¹³C, ⁷⁷Se NMR spectra, and FAB-HRMS¹³ and by comparison with the 7β-epimer, which was independently obtained through the coupling procedure by using only one equivalent of sodium hexamethyldisilazide.

For the preparation of the 5-membered C-ring taxol analog (10), taxol (7a) was treated with t-butyldimethylsilyl chloride and imidazole in dimethyl formamide 14 to give 2'-(t-butyldimethylsilyl)taxol (7b)

Scheme I

(a) Sml₂, HOAc, THF; (b) SeO₂ Pyr; (c) NaHMDS, Ojima's β-lactam (5), THF; (d) HF, Pyr

in 99% yield (Scheme II). Treatment of **7b** with trifluoromethanesulfonic anhydride and pyridine in dichloromethane at 25 °C and subsequently with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) at the same temperature for 20 h afforded 7-deoxy-2'-(t-butyldimethylsilyl)-Δ^{6,7}-taxol (**8a**) in 70% yield together with 24% of the deprotected compound **8b**. Dihydroxylation of **8a** with osmium tetroxide in pyridine-THF (1:1) at 25 °C yielded 2'-(t-butyldimethylsilyl)-6α-hydroxy-7-epitaxol (**9**) in 85% yield. ¹⁵ Cleavage of diol **9** with lead tetraacetate in the presence of sodium bicarbonate in benzene at 25 °C yielded three compounds.

Scheme II

The major component was isolated in 40% yield after deprotection of the t-butyldimethylsilyl group. Extensive spectroscopic characterization of this compound 16 led to the structural assignment as the C ring-contracted taxol 10.

Both structurally modified taxol analogs 6 and 10 were directly compared to taxol for their biological activities. At identical concentrations, both analogs were found to promote polymerization of tubulin and stabilize the microtubules, although neither was as active as taxol. Both analogs are also less cytotoxic than taxol for J774.2 cells. ED₅₀ values for taxol and analogs 6 and 10 are shown in Table I. These results show that activity is retained in A- and C-ring modified taxol analogs 6 and 10. Both analogs can mimic the active

binding conformation of taxol. Conformational analyses of 6 and 10 by NMR spectroscopy and molecular modeling indicated that both assume conformations similar to taxol in chloroform.¹⁷

Table I: Cytotoxicity of taxol analogs 6 and 10 versus taxol

compound	ED so(µM)
taxol	0.016
6	0.90
10	0.25

Exponentially growing J774.2 cells (2 x 10⁴ / mL) were placed in multi-well plates. Either taxol, 6, or 10 was added at various concentrations and incubated at 37°C. After 72h the cell number was determined.

Previous work has shown that taxol-like activity is retained in taxol analogs with partially modified or deleted functionality. The current and related studies suggest that activity can also be retained through modifications of the taxol carbotricyclic core. Further studies are in progress.

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References and Notes

- 1. Taxol is a registered trademark of Bristol-Myers Squibb Inc. which offers the generic name paclitaxel.
- Wani, M. C.; Taylor, H. L.; Wall, M. E.; Coggen, P.; McPhail, A. T. J. Am. Chem. Soc. 1971, 93, 2325.
- (a) Rowinsky, E. K.; Cazenave, L. A.; Donehower, R. C. J. Natl. Cancer Inst. 1990, 82, 1247; (b) Slichenmyer, W. J.; Von Hoff, D. D. J. Clin. Pharmacol. 1990, 30, 770; (c) Slichenmyer, W. J.; Von Hoff, D. D. Anti Cancer Drugs 1991, 2, 519.
- 4. Suffness, M. Ann. Rep. Med. Chem. 1993, 28, 305.
- (a) Schiff, P. B.; Fant, J.; Horwitz, S. B. Nature 1979, 277, 665; (b) Horwitz, S. B. Trends Pharmacol. Sci. 1992, 13, 134; (c) Andreu, J. M.; Diaz, J. F.; Garcia de Ancos, J.; Gil, R.; Mendrano, F. J.; Nogales, E.; Pantos, E.; Towns-Andrews, E. J. Mol. Biol. 1992, 226, 169; (d) Diaz, J. F.; Andreu, J. M. Biochemistry 1993, 32, 2747; (e) Jordan, M. A.; Toso, R. J.; Thrower, D.; Wilson, L. Proc. Natl. Acad. Sci. U. S. A. 1993, 90, 9552; (f) Arnal, I.; Wade, R. H. Current Biol. 1995, 5, 900; (g) ter Har, E.; Kowalski, R. J.; Hamel, E.; Lin, C. M.; Longley, R. E.; Gunasekera, S. P.; Rosenkranz, H. S.; Day, B. W. Biochemistry 1996, 35, 243; (h) Hung, D. T.; Chen, J.; Schreiber, S. L. Chem. Biol. 1996, 3, 287.
- (a) Suffness, M.; Cordell, D. A. The Alkaloids: Chemistry and Pharmacology; Academic: New York, 1985, Vol. 25, pp 3–355; (b) Blechert, S.; Guénard, D. The Alkaloids: Chemistry and Pharmacology; Academic: San Diego, 1990; Vol. 39, pp 195–238.
- Academic: San Diego, 1990; Vol. 39, pp 195-238.
 (a) Rao, S.; Horwitz, S. B.; Ringel, I. J. Natl. Cancer Inst. 1992, 84, 785; (b) Rao, S.; Krauss, N. E.; Heerding, J. M.; Swindell, C. S.; Ringel, I.; Orr, G. A.; Horwitz, S. B. J. Biol. Chem. 1994, 269, 3132; (c) Rao, S.; Orr, G. A.; Chaudhary, A. G.; Kingston D. G. I.; Horwitz, S. B. J. Biol. Chem. 1995, 270, 20235; (d) Dasgupta, D.; Park, H.; Harriman, G. C. B.; Georg, G. I.; Himes, R. J. Med. Chem. 1994, 37, 2976.
- 8. (a) Wender, P. A.; Mucciaro, T. P. J. Am. Chem. Soc. 1992, 114, 5878; (b) Wender, P. A.; Badham, N. F.; Conway, S. P.; Floreancig, P. E.; Glass, T. E.; Gränicher, C.; Houze, J. B.; Jänichen, J.; Lee, D.; Marquess, D. G.; McGrane, P. L.; Meng, W.; Mucciaro, T. P.; Mühlebach, M.; Natchus, M. G.; Paulsen, P.; Rawlins, D. B.; Satkofsky, J.; Shuker, A. J.; Sutton, J. C.; Taylor, R. E.; Tomooka, K. J. Am. Chem. Soc. 1997, 119, 2755; (c) Wender, P. A.; Badham, N. F.; Conway, S. P.; Floreancig, P. E.; Glass, T. E.; Houze, J. B.; Krauss, N. E.; Lee, D.; Marquess, D. G.; McGrane, P. L.; Meng, W.; Natchus, M. G.; Shuker, A. J.; Sutton, J. C.; Taylor, R. E. J. Am. Chem. Soc. 1997, 119, 2757; (d) Georg, G. I.; Chen, T. T.; Ojima, I.; Vyas, D. M. Taxane Anticancer Agent: Basic Science and Current Status; American Chemical Society: Washington, DC, 1994; (d) Suffness, M. Taxol®: Science and Applications; CRC: Boca Raton, FL, 1995.
- 9. During the preparation of this manuscript, the synthesis of 10 was published. (a) Liang, X.; Kingston, D. G. I.; Long, B. H.; Fairchild, C. A.; Johnston, K. A. Tetrahedron Lett. 1995, 36, 7795; (b) Liang, X.; Kingston, D. G. I.; Long, B. H.; Fairchild, C. A.; Johnston, K. A. Tetrahedron 1997, 53, 3441.
- 10. Georg, G. I.; Cheruvallath, Z. S. J. Org. Chem. 1994, 59, 4015.

- A similar skeletal rearrangement was observed on a related systems. (a) Samaranayake, G.; Magri, N. F.; Jitrangsri, C.; Kingston, D. G. I. J. Org. Chem. 1991, 56, 5114; (b) Chen, S.-H.; Huang, S.; Wei, J.; Farina, V. J. Org. Chem. 1993, 58, 4520; (c) Zamir, L.; Zheng, Y. F.; Caron, G.; Sauriol, F.; Mamer, O. Tetrahedron Lett. 1996, 37, 6435. Spectroscopic data for 3: 1H NMR (CDCl₃, 400 MHz) δ 8.19 (t, J = 46 Hz, 1H, 18-H), 8.08 (d, J = 7.2 Hz, 2H), 7.64 (t, J = 7.5 Hz, 1H), 7.49 (t, J = 7= 7.6 Hz, 2H), 6.25 (d, J = 8.9 Hz, 2H), 5.18 (q, J = 6.2 Hz, 1H, 13-H), 4.92 (dd, J = 9.2, 2.8 Hz, 1H, 5-H), 4.55 (d, J = 8.2 Hz, 1H, 20-H), 4.18 (d, J = 8.2 Hz, 1H, 20-H), 3.76 (ddd, J =10.1, 6.9, 1.5 Hz, 1H, 7-H), 3.18 (d, J = 9.5 Hz, 1H, 3-H), 3.08 (d, J = 1.5 Hz, 1H, 7-OH), 3.00 (dd, J = 14.3, 6.7 Hz, 1H, 14-H), 2.58 (ddd, J = 14.4, 9.2, 6.9 Hz, 1H, 6-H), 2.46 (dd, J = 14.3, 7.3 Hz, 1H, 14-H), 2.18 (s, 3H, 4-OAc), 2.06 (d, J = 6.5 Hz, 1H, 13-OH), 1.94 (ddd, J = 14.4, 10.1, 2.8 Hz, 1H, 6-H), 1.84 (s, 3H, 19-Me), 1.11 (s, 3H, 17-Me), 1.02 (s, 3H, 16-Me). The selenium NMR spectra was recorded using neat PhSeH as external standard. ⁷⁷Se NMR (CDCl₃, 95.3 MHz) δ 1959.68 (d, J = 46 Hz). ¹³C NMR (CDCl₃, 100 MHz) δ 203.9, 169.8, 165.9, 154.5, 153.8, 133.8, 129.8, 129.5, 129.3, 128.7, 83.7, 79.9, 75.5, 75.2, 69.9, 69.7, 69.2, 62.8, 57.4, 46.9, 45.6, 35.5, 27.0, 26.0, 21.8, 9.6. HRMS (FAB) calcd for $C_{29}H_{33}O_{9}^{80}Se$ (MH+) 605.1289, found 605.1286. The structure of 3 was further secured by x-ray crystallography. Crystallographic data (collected at CHEXRAY of UC Berkeley): A colorless orthorhombic crystal with dimensions of 0.14 x 0.25 x 0.30 mm³ was mounted on a glass fiber in paratone oil at -80 °C. The space group is C222₁ (#20) and cell constants are a = 10.0936(2) Å, b = 15.3218(2) Å, c = 35.5927(4) Å, V = 5504.5(1) Å³, and Z = 8. All measurements were made on a Siemens SMART diffractometer with graphitemonochromated Mo-K α radiation. The data were collected at a temperature of -129 \pm 1°C using the ω scan technique to a maximum 2θ value of 46.6°. The structure was solved by direct method (SHELXS-86) and expanded using Fourier techniques. The final cycle of full-matrix least-square refinement was based on 3422 observed reflections $(I > 3\sigma(I))$ and 361 variable parameters and converged with R = 0.030, $R_{w} = 0.035$.
- (a) Ojima, I.; Habus, I.; Zhao, M.; Georg, G. I.; Jayasinghe, L. R. J. Org. Chem. 1991, 56, 1681;
 (b) Holton, R. A. Chem. Abstr. 1990, 114, 164568q.
- 13. Spectroscopic data for 6: 1 H NMR (CDCl₃, 400 MHz) δ 8.20 (d, J = 7.8 Hz, 2H), 8.01 (s, 1H, 18-H), 7.70 (d, J = 7.8 Hz, 2H), 7.57 (t, J = 7.5 Hz, 1H), 7.34-7.52 (m, 10H), 6.93 (d, J = 9.2 Hz, 3'-NH), 6.28 (d, J = 9.5 Hz, 1H, 2-H), 6.11 (t, J = 7.0 Hz, 1H, 13-H), 5.79 (dd, J = 9.2, 1.8 Hz, 1H, 3'-H), 4.93 (t, J = 6.4 Hz, 1H, 5-H), 4.75 (dt, J = 3.6, 1.8 Hz, 1H, 2'-H), 4.68 (d, J = 8.5 Hz, 1H, 20-H), 4.25 (d, J = 8.5 Hz, 1H, 20-H), 4.15 (dt, J = 11.8, 3.5 Hz, 1H, 7-H), 4.02 (d, J = 11.8 Hz, 1H, 7-OH), 3.44 (d, J = 3.6 Hz, 1H, 2'-OH), 3.40 (d, J = 9.5 Hz, 1H, 3-H), 2.91 (d, J = 7.0 Hz, 2H, 14-H), 2.38 (s, 3H, Me), 2.36 (dd, J = 6.4, 3.5 Hz, 2H, 6-H), 1.78 (s, 3H, Me), 1.11 (s, 3H, Me), 77Se NMR (CDCl₃, 95.3 MHz) δ 1948.92 (d, J = 46 Hz). 13 C NMR (CDCl₃, 100 MHz) δ 202.6, 172.6, 172.0, 165.9, 151.6, 148.0, 138.3, 138.1, 133.9, 133.7, 131.8, 130.5, 130.2, 129.3, 129.0, 128.8, 128.6, 128.1, 127.0, 126.9, 126.8, 83.0, 81.1, 76.1, 75.7, 73.6, 73.3, 68.9, 63.0, 56.6, 54.7, 42.2, 41.5, 35.5, 29.7, 26.9, 25.9, 21.8, 15.8. HRMS (FAB) calcd for C₄₅H₄₆NO₁₂80Se (MH+) 872.2185, found 872.2235.
- 14. Magri, N. F.; Kingston, D. G. I. J. Nat. Prod. 1988, 51, 298.
- (a) Johnson, R. A.; Nidy, E. G.; Dobrowolski, P. J.; Gebhard, I.; Qualls, S. J.; Wicnienski, N. A.; Kelly, R. C. Tetrahedron Lett. 1994, 35, 7893; (b) Liang, X.; Kingston, D. G. I.; Lin, C. M.; Hamel, E. Tetrahedron Lett. 1995, 36, 2901.
- 16. Spectroscopic data for 10: 1 H NMR (CDCl₃, 400 MHz) δ 8.20 (d, J = 8.8 Hz 2H), 7.66 (d, J = 8.5 Hz, 2H), 7.61 (t, J = 7.6 Hz, 1H), 7.31-7.53 (m, 10H), 6.95 (d, J = 9.1 Hz, 3'-NH), 6.45 (s, 1H, 10-H), 6.27 (br t, J = 8.5 Hz, 1H, 13-H), 5.81 (m, J = 9.1 Hz, J = 7.6, 1.5 Hz, 2H, 2-H, 3'-H), 5.13 (d, J = 4.4 Hz, 1H, 5-H), 4.79 (d, J = 1.5 Hz, 1H, 2'-H), 4.45 (d, J = 8.2 Hz, 1H, 20-H), 4.35 (d, J = 7.6 Hz, 1H, 3-H), 4.33 (dd, J = 7.6, 4.4 Hz, 1H, 7-H), 4.28 (d, J = 8.2 Hz, 1H, 20-H), 3.48 (br s, 1H, 2'-OH), 2.53 (d, J = 7.6 Hz, 1H, 7-OH), 2.43 (s, 3H, Me), 2.32 (dd, J = 15.6, 9.5 Hz, 1H, 14-H), 2.21 (s, 3H, Me), 2.20 (dd, J = 15.6, 8.5 Hz, 1H, 14-H), 1.85 (s, 3H, Me), 1.84 (s, 3H, Me), 1.23 (s, 3H, Me), 1.14 (s, Me, 3H). 13 C NMR (CDCl₃, 100 MHz) δ 204.3, 172.6, 169.7, 167.0, 166.9, 140.4, 138.0, 135.9, 133.7, 133.5, 131.9, 130.3, 129.2, 128.8, 128.7, 128.6, 128.3, 127.0, 126.8, 87.7, 81.7, 78.6, 75.6, 74.2, 73.5, 73.2, 72.3, 59.8, 54.7, 44.5, 42.7, 36.9, 26.5, 22.0, 21.0, 20.7, 14.6, 12.9. HRMS (FAB) calcd for C₄₆H₄₉NNaO₁₄ (MNa+) 862.3051, found 862.3063.
- (a) Williams, H. J.; Scott, A. I.; Dieden, R. A.; Swindell, C. S.; Chirlian, L. E.; Francl, M. M.; Heerding, J. M.; Krauss, N. E. Can. J. Chem. 1994, 72, 252; (b) Dubois, J.; Guénard, D.; Guéritte-Voegelein, F.; Guedira, N.; Poiter, P.; Gillet, B.; Beloeil, J-C. Tetrahedron 1993, 49, 6533; (c) Paloma, L. G.; Guy, R. K.; Wrasidlo, W.; Nicolaou, K. C. Chem. Biol. 1994, 1, 107.